

Al-Nb-Si (Aluminum-Niobium-Silicon)

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Recently, [2004Sha] presented a thermodynamic assessment of this system and computed a liquidus projection and two isothermal sections at 1400 and 1000 °C. The available experimental data on the isothermal sections show good agreement with the computed sections.

Binary Systems

The Al-Nb phase diagram [Massalski2] has three intermediate phases: NbAl₃ (D0₂₂, TiAl₃-type tetragonal), Nb₂Al (30-42 at.% Al; D8_b, σCrFe-type tetragonal), and Nb₃Al (18.6-25 at.% Al; A15, Cr₃Si-type cubic). The Al-Si

system is of the simple eutectic type, with the eutectic temperature at 577 °C and the composition at 12.2 at.% Si. The Nb-Si phase diagram [1993Sch, 1995Sch] has the following intermediate phases. Nb₃Si (Ti₃P-type tetragonal), βNb₅Si₃ (D8_m, W₅Si₃-type tetragonal), αNb₅Si₃ (D8₁, Cr₅Si₃-type tetragonal), and NbSi₂ (C40, CrSi₂-type hexagonal).

Ternary Phases

Two ternary phases have been identified in this system [1961Bru]. Al₂Nb₃Si₅ (denoted τ₁ here) has the C54, TiSi₂-

Table 1 Al-Nb-Si crystal structure and lattice parameter data

Phase	Composition, at. %	Pearson symbol	Space group	Prototype	Lattice parameter, nm
Al ₂ Nb ₃ Si ₅ (τ ₁)	~20 Al ~30 Nb ~50 Si	<i>oF</i> 24	<i>Fddd</i>	TiSi ₂	<i>a</i> = 0.8403 <i>b</i> = 0.4901 <i>c</i> = 0.8794
Al ₃ Nb ₁₀ Si ₃ (τ ₂)	~18.75 Al ~62.5 Nb ~18.75 Si	<i>tI</i> 32	<i>I4/mcm</i>	W ₅ Si ₃	<i>a</i> = 1.016 <i>c</i> = 0.508

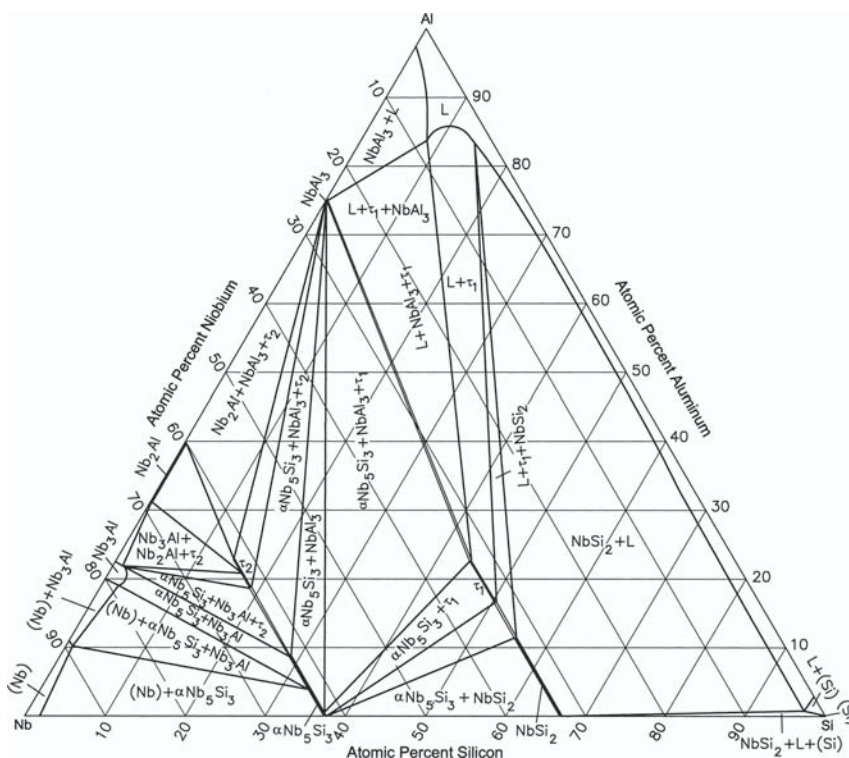


Fig. 1 Al-Nb-Si computed isothermal section at 1400 °C [2004Sha]

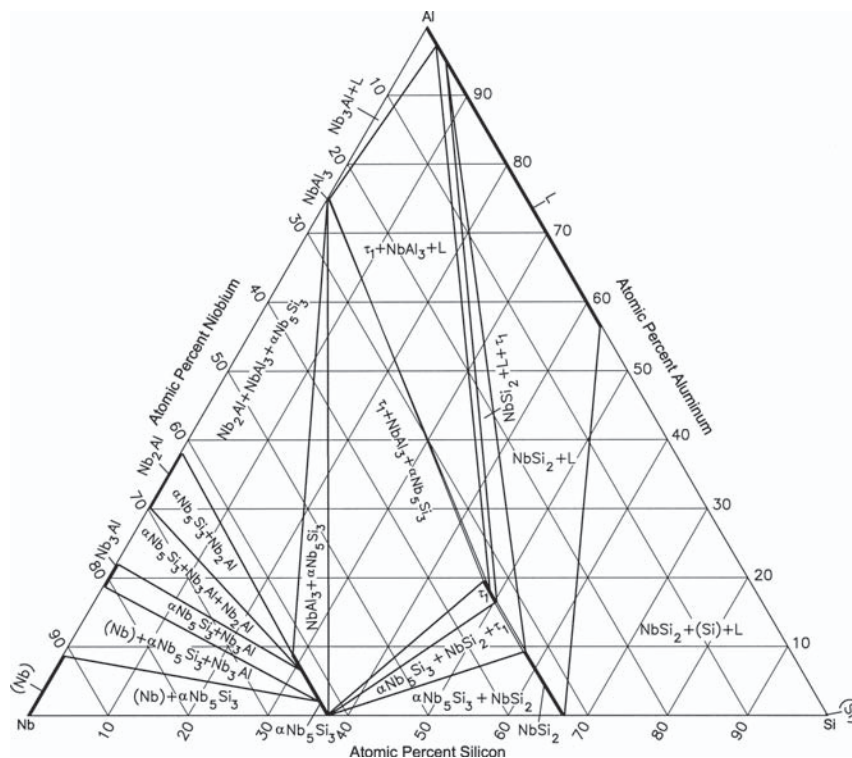


Fig. 2 Al-Nb-Si computed isothermal section at 1000 °C [2004Sha]

type orthorhombic structure, with a small homogeneity range at constant Nb content. The other compound $\text{Al}_3\text{Nb}_{10}\text{Si}_3$ (denoted τ_2 here and β by [2004Sha]) has the $D8_m$ -type tetragonal structure of $\beta\text{Nb}_5\text{Si}_3$. It is apparently stabilized by Al substituting for Si. It is stable at 1400 °C as a ternary phase, even though the decomposition temperature of $\beta\text{Nb}_5\text{Si}_3$ in the binary system is 1650 °C. τ_2 was not found at 1000 °C by [2003Zha].

Isothermal Sections

The experimental investigations of this system are [1961Bru] (a composite isothermal section at 1400 °C for Nb-rich alloys and at 500 °C for Nb-poor alloys), [1973All] (partial isothermal sections for Nb-poor alloys at 1500 and 1300 °C), [1984Pan] (partial isothermal section at 1500 °C for Nb-rich alloys), [2001Mur] (isothermal section valid between 1600-1200 °C), and [2003Zha] (isothermal section at 1000 °C). For the isothermal sections of [1961Bru], [1973All], and [1984Pan], see [1995Vil]. [2001Mur] used powders of 99.9% Al, 99.8% Nb, and 99.9% Si to prepare samples by compacting and sintering in the temperature range of 1200-1600 °C for 0.3-0.6 ks. They confirmed the presence of the two ternary compounds τ_1 and τ_2 and drew a tentative isothermal section valid for the above temperature range. Crystallographic data for these two ternary phases are given in Table 1. Using a diffusion-multiple approach, [2003Zha] constructed an isothermal section at 1000 °C, which depicts the ternary phase τ_1 . The Al-stabilized

$\beta\text{Nb}_5\text{Si}_3$ -based ternary phase (τ_2) was not found at this temperature. Tie-lines form between $\alpha\text{Nb}_5\text{Si}_3$ and NbAl_3 , confirming the results of [1984Pan] and [2001Mur]. The Al solubility in $\alpha\text{Nb}_5\text{Si}_3$ is ~8 at.%, with the Al atoms substituting for Si.

[2004Sha] developed a thermodynamic description of this system using the CALPHAD approach. For the Al-Nb and Al-Si systems, the earlier descriptions from the literature were used. The Nb-Si system was reoptimized by [2004Sha], using a term for the excess entropy of mixing of the liquid phase. The computed isothermal sections at 1400 and 1000 °C are redrawn in Fig. 1 and 2. [2004Sha] found good agreement with the experimental data of [1961Bru], [2001Mur], and [2003Zha]. [2004Sha] computed a liquidus projection, which depicts that τ_1 and $\alpha\text{Nb}_5\text{Si}_3$ form peritectically in the ternary region. There are no experimental data to compare with the computed projection.

References

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